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Ionic bonding is when two ions of opposite charge attract and bond to each other forming an ionic compound. Consider the following examples of formulas for ionic compounds: • One  $\text{Na}^+$  (sodium ion) and one  $\text{Cl}^-$  (chloride ion) bond to make  $\text{NaCl}$ , “sodium chloride.” • One  $\text{Mg}^{2+}$  (magnesium ion) and two  $\text{F}^-$  (fluoride ion) bond to make  $\text{MgF}_2$

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Ionic bond, also called electrovalent bond, type of linkage formed from the electrostatic attraction between oppositely charged ions in a chemical compound. Such a bond forms when the valence (outermost) electrons of one atom are transferred permanently to another atom.

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