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Directed Reading A Section: Energy and Rates of Chemical Reactions 1. All chemical reactions either give off or absorb _____. REACTIONS AND ENERGY 2. Why is chemical energy a part of all chemical reactions? _____ 3. When energy is released during a chemical reaction, it is called a(n) _____ reaction. 4.

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Similarly, chemical energy is the energy stored in molecular bonds. In a chemical reaction bonds are broken and formed. Breaking bonds requires energy, making bonds produces energy. Types of energy include radiant, thermal, kinetic and potential energy. When charged particles move they emit electromagnetic radiation which we call radiant energy. This also includes quantum state changes in atoms and molecules which result in the emission/absorption of light.

Chapter 15.1 Energy Changes in Chemical Reactions ...

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Section: Energy and Rates of Chemical Reactions 1. All chemical reactions either give off or absorb _____.

REACTIONS AND ENERGY 2. Why is chemical energy a part of all chemical reactions?

Skills Worksheet Directed Reading A use with pages 42 - 47.

type of chemical reaction in which bonds are both formed and broken, resulting in the transfer of components kinetic energy energy that matter possesses because of its motion potential energy stored energy matter possesses because of the positioning or structure of its components product one or more substances produced by a chemical reaction reactant

2.3 Chemical Reactions | Anatomy and Physiology

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Formulas and Equations ... A substance that forms in a chemical reaction is called a(n) . 18. ... In reactions, energy is released. 5. A solid that is produced as a result of a chemical reaction in solution is a(n) .

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The smallest amount of energy needed to start a chemical reaction is called _____. Activation energy. Name sources

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chemical reaction that releases energy
(release) endergonic reactions a
chemical reaction that requires more
energy to break bonds than is released
when new ones are formed (take in)

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