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A. 1.20×10^{23} B. 3.01×10^{23} C. 6.02×10^{23} D. 1.20×10^{24}

2. 1 mol of a hydrocarbon with general formula C_nH_{2n+2} reacts completely with oxygen to produce 4 mol of H_2O . What is the amount of oxygen molecules, in mol,

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that reacts? A. 4 B. 5 C. 6 D. 7 3. Under which combination of conditions is 1 mol of an ideal gas present? Volume Pressure ...

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The molar mass of a compound is approximately 56 g mol⁻¹. Which

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formula is possible for this Chemistry SL
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When 100 cm³ of 1.0 mol dm⁻³ HCl is
mixed with 100 cm³ of 1.0 mol dm⁻³
NaOH, the temperature of the resulting
solution increases by 5.0 ° C.

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