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ChemActivity 50 The Electrochemical Cell 293 $\text{Cu}^{2+}(\text{1 M}) + 2\text{e}^- \text{Cu}(\text{s})$ Simultaneously, at the Pt electrode (anode), the following reaction takes place: $\text{H}_2(\text{g}; 1 \text{ atm}) \rightarrow 2\text{H}^+(\text{1 M}) + 2\text{e}^-$ The experimental voltage, E° , is 0.34 V. Critical Thinking Questions 11.

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PLEASE NOTE: If you have a question about these answers, it is your responsibility to come to office hours or ask during class work time. UNIT 12 - HW Practice Keys - ChemActivity 51: Cell Voltage - ChemActivity 50: Electrochemical Cell UNIT 11 - HW Practice Keys - ChemQuest 55: Free Energy - ChemQuest 54: 2nd Law of Thermodynamics

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9. Which is more concentrated with respect to sodium ions, 0.50 M NaCl or 0.30 M Na₂SO₄? 10. Which CTQ was easier to answer, 8 or 9? Why? 182 ChemActivity 32 Molarity Information Some molecules do not dissociate into ions when dissolved in water. Sugars (glucose, sucrose, dextrose, etc.) and alcohols are examples.

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Electrochemical Cells . ChemActivity 50: CTQ (1-13) ChemActivity 50(a): CTQ (1-3) W. Nov 16 . Electrolytic Cells . ChemActivity 51: CTQ (1-4) Ex. 1, 2 Pr. 1, 3, 4. ChemActivity 51(a): CTQ(1-7) F. Nov 18 . Balancing Redox Reactions . ChemActivity 49: CTQ (1-7) Ex. 1, 5 ...

Chem 124 Homework

ChemActivity Colligative Properties 3. Calculate the mass of the solvent added to the solution in model 1 (in kg). 4. What is the molality of the KCl solution in model 1? 5. Why are your answers to CTQ 3 and 4 different? 6. Which has the greater molality, 50.0 g of NaCl dissolved in 100.0 ml of water or 50.0 g of NaCl dissolved in 1000 ml of ...

Solved: ChemActivity Colligative Properties ChemActivit Co ...

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